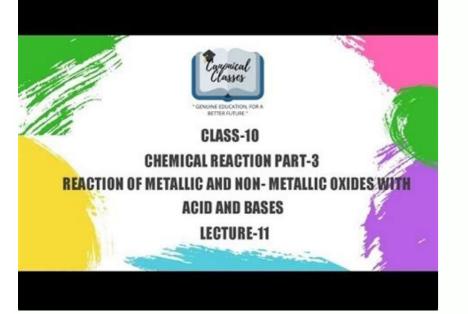
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ACBSE Coaching for Mathematics and Science

- 9. If the weight of 6 sheets of a paper is 162 grams, how many sheets of the same quality of paper would weigh 13.5 kg?
- 6:0.162 = x:13.5
- 10. 1152 bars of soap can be packed in 8 cartons of the same size. How many such cartons will be required to pack 3888 bars?
- 1152 : 8 = 3888: x

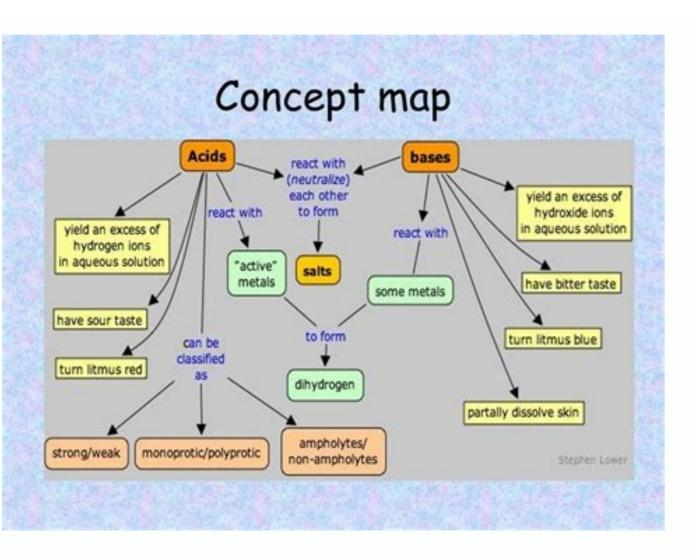
15: 16.5 = x : 26

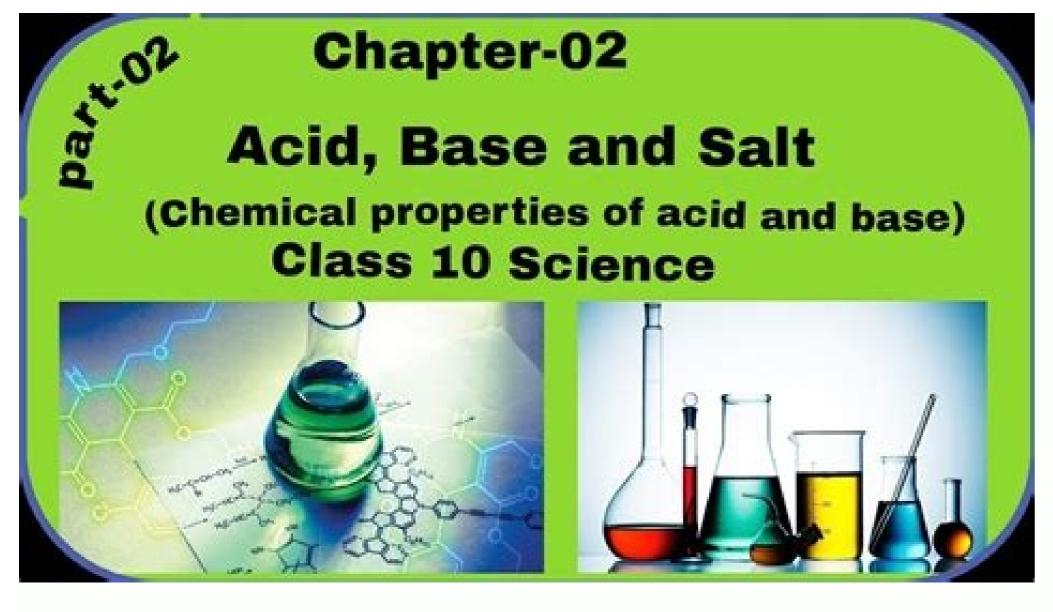
- 11. If the thickness of a pile of 16 cardboards is 44 mm, how many cardboards will be there in a pile which is 71.5 cm thick?
- 16:4.4 = x:71.5
- 12. At a particular time of day, a 7m high flagstaff casts a shadow which is 8.2m long. What is the height of the building which casts a shadow 20.5 meters in length at the same moment?
- 7:8.2:x:20.5

 13. 15 men can build a 16.25 m long wall up to a certain height in one day. How many men should be
- employed to build a wall of the same height but of length 26 m in one day?
- 14. In a hospital, the monthly consumption of milk of 60 patients is 135 liters. How many patients can be
- accommodated in the hospital if the monthly ration of milk is raised to 1710 liters, assuming that the quota per head remains the same?
- 60: 135 = x: 1710 => $x = \frac{60 \times 1710}{135} = 76 \ days$
- 15. The extension in an elastic string varies directly as the weight hung on it. If a weight of 150 g produces an extension of 2.8 cm, what weight would produce an extension of 19.6 cm?
- $0.150: 2.8 = x: 19.6 \implies x = \frac{0.150 \cdot x \cdot 19.6}{2.8} = 1 \text{ kg 50g}$
- 16. If 13 metres of a uniform iron rod weighs 23.4 kg then what will be the weight of 6 metres of the same rod?

 13: 23.4 = 6: x
- 17. The length of the shadow of a 3-m-high pole at a certain time of the day is 3.6 m. What is the height of another pole whose shadow at that time is 54 m long?

 3:3.6 = x:54
- 18.If the cost of 9 toys is Rs 333, find the cost of 16 such toys. 9:333 = 16:x
- 19.If 25 metres of cloth costs Rs 1575, how many metres of it can be bought for Rs 2016?
- 25 : 1575 = x : 2016
- Document Downloaded from https://isuniltutorial.weebly.com (RS Agarwal EXERCISE 9) Page | 2





Acid base or salt worksheet answer key. Class 10 chapter acid base and salt 100 question with answer. Acid base and salt class 10 short notes. Acid base and salts class 10 worksheet.

In this page we have Class 10 Science Acid base and Salts Practice worksheet. Hope you like them and do not forget to like, social shar and comment at the end of the page. Question 1 Name four gases that can be prepared in the laboratory using dilute H2SO4. Show how they can be prepared? Solution The four gases are: 1) H2 gas: An active metal (above Pb) and dilute H2SO4 gives H2 gas 2) CO2 gas :Any carbonate salt and dilute H2SO4 gives CO2 gas 3) SO2 gas :Any sulphite salt and dilute H2SO4 gives H2S gas. Question 2 atch the column Column A Column B HCL Salt NaOH Acid HNO3 Base NH4OH CuCO3 (NH4)2SO4 Solution Acid:HCL,HNO3 Base:NaOH,NH4OH Salt:CuCO3,NH4)2SO4 Question 3 A milkman adds a very small amount of baking soda to fresh milk take a long time to set as curd? Solution So that it does not set as curd easily Since Milk is alkaline in nature, acid produced are neutralized by the base, so it takes longer time to set into curd Question 4 Fill in the blanks. (a) Chemical formula for bleaching powder is and water. (c) The term pH was coined by bases to form medicine used to treat indigestion (g) Calcium Sulphate hemihydrate is popularly known as are present in aerated drinks Solution a)CaOCl2 b)React, with. salt d) H+ e)Washing Soda f)antacid g) Plaster of Paris i) CO2 Question 5 atch the column Column A Column B Common Salt NaOH Baking soda Na2CO3 Caustic soda NaCl Washing Soda: NaCl Washing Soda: NaCl Washing Soda: NaHCO3 Caustic soda: NaOH Washing Soda: NaUcl Backing soda: NaHCO3 Caustic soda: NaCl Washing Soda: NaUcl Backing soda: NaHCO3 Caustic soda: NaOH Washing Soda: NaUcl Backing sod a solution can vary from 0-14. f) Heat is evolved during ionization. g) NaCl is a base h) Acids turns blue litmus solution to red Solution a) True b) True c) True d) True b) True do not storage and a strong base and a strong bas weak base. e) Name a hydrated salt f) What is a universal indicator? What is its advantage? Solution a) Limewater is the common name for a diluted solution of calcium hydroxide c)Range 1-3 d)Strong Base is N4OH e) A hydrated salt is copper sulphate crystals (CuSO4.5H2O). f) A universal indicator is a mixed indicator of organic chemicals which not only shows whether the given solution is acidic or basic, but also shows the approximate pH values by giving a wide particular colour for a specific value of pH. link to this page by copying the following textClass 10 Science Acid base and Salts Practice worksheet Also Read Practice Question Class 10 Maths Class 10 ions.2. Higher the pH value of a solution, the more Science Fill in the blanks with suitable words1. An acid is a compound which when dissolved in water gives ions as the only it is.3. pH value of lemon juice is than 7.4. Salts of strong acids and strong bases in water produce acid.7. Dissolving of aluminium foil in HCl as well as NaOH shows that it is in nature.8. Methyl orange turns gas is evolved when dil. HCl reacts with metal bicarbonate.12. Tooth decay start when the pH of mouth is lower than _____.13. Fresh milk has a pH of 6. When it changes into curd (yoghurt), its pH value acid because it gets completely dissociated to liberate the H+ ions.11. is the fixed number of water molecules present in one formula unit of a substance.17. When a metallic oxide is dissolved in water, the solution formed has high concentration of chemical formula of the calcium compound used for disinfecting drinking water is .15. Plaster of Paris is prepared by decomposing ions.18. The salt of the sulphuric acid are called sulphates and those of the acid HNO3 are called is the process of decreasing the strength of an acid or base by hydrolysis.20. The pH range our body works is _____. Answers:1. Hydronium or Hydrogen ion, Positive2. alkaline / basic3. less 4. neutral5. basic, acid6. acetic7. amphoteric8. pink9. neutralisation10. strong11. Carbon Dioxide12. 5.513. decreases (due to formation of more lactic acid)14. Calcium oxychloride (CaOCl2)15. gypsum16. Water of crystallization17. OH-18. nitrates19. Dilution20. 7.0-7.8See Also:Ch2 - Acids, Bases and Salts (MCQs, NCERT Solution, Q & A)Ch2 - Acids, Bases and Salts (MCQs)Ch2 - Acids, Bases and Salts (M Acids, Bases and Salts (Worksheet)Ch 2 - Acids, Bases and Salts (Very Short Q A)Ch2 - Acids, Bases and Salts - Indicators (Q & A)Ch2 - Acids, Bases and Salts - Acids, Bases and Salts - Acids, Bases & Salts (Assert-Reason Based Questions - with Answers) Colour of Compounds/Salts Read and download free pdf of CBSE Class 10 Science can get free printable Worksheets for Class 10 Science in PDF format prepared as per the latest syllabus and examination pattern in your schools. Standard 10 students should practice questions and answers given here for Science in Grade 10 which will help them to improve your knowledge of all important chapters and its topics. Students should also download free pdf of Class 10 Science Worksheets prepared by school teachers as per the latest NCERT, CBSE, KVS books and syllabus issued this academic year and solve important problems provided here with solutions on daily basis to get more score in school exams and testsAcids Bases And Salts in standard 10. This test paper with questions and answers for Grade 10 Science Worksheet for Acids Bases And SaltsMCQ Questions for NCERT Class 10 Science Worksheet for Acids Bases And SaltsQuestion on pH strip, the colour of pH strip changed to green. The liquid taken is likely to bea. pH paper shows green colour with water us neutral.b. dilute sodium hydroxide solutions. Higher the hydronium ion concentration, a. higher is the pH valuec. moderate is the pH valueExplanation: The hydrogen ion concentration is represented by the symbol[H+]. The pH is defined as pH = -log [H+] ie, it's the negative logarithm of the H+ concentration. In water H+ is really H3O+ (the hydronium ion). The greater the concentration is pH is defined as pH = -log [H+] ie, it's the negative logarithm of the hydronium ion, the lower the value of pH. It is simply logarithms. It has nothing to do with the chemistry behind it. Question: carbonates and hydrogen carbonates are types of base. They make a salt and water when we neutralise them with acid we get carbon dioxide are given off. This is easy to remember because we see the word 'carbonate' in the chemical names.acid + metal hydrogen carbonate → a salt + water + carbon dioxideQuestion : On adding a drop of given solution was b. dilute solution of sodium bicarbonatec. dilute hydrochloric acidd. concentration hydrochloric acidExplanation: Dilute solution of NaHCO3 is basic in nature. It will turn pH paper blue. Question: Raw materials for baking soda and tartaric acidExplanation: Baking powder is a solid mixture that is used as a chemicalleavening agent in baked goods. Baking powder is a mixture of sodiumhydrogen carbonate and weak edible acid named as tartaric acid. Question: Which of the following salts contains water of crystallization? Explanation: All the above salts contain water of crystallization and their chemical formulae are given below:Gypsum - CaSO4.2H2O - (2 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - CuSO4.5H2O - (5 molecules of water of crystallisation) Flus vitriol - C gas.Answer: hydrogenQuestion. The salts of a strong acid and weak base are with pH value than 7.Answer: acidic, lessQuestion. During indigestion. He stomach products obtained when bleaching powder reacts with dilute sulphuric acid. Answer: CaSO4, Cl2, H2OQuestion. The aqueous solution of an acid or a base conducts electricity due to the presence of in it.Answer: ionsQuestion. A base which is soluble in water is known as an .Answer : alkaliQuestion. Among HCl, H2SO4 and CH3COOH, is a weak acid.Answer ... is the fixed number of water molecules chemically attached to each formula unit of a salt in its crystalline form. Answer: Water of crystalline form. Answer: Universal indicator. Question. Which acid is present in honey bee sting and hair of nettle leaves? Answer: Formic acidQuestion. What effect does the concentration of H+(ag) have on the acidic nature of the solution? Answer: Acidic nature increases with increase in cone, of H3O+ ion. Question. Name flowers whose colour or petals can be used as indicator. Answer: Hydrangea, Petunia GeraniumQuestion. Define universal indicator. Answer: A universal indicator is a mixture of indicators which shows a gradual but well marked series of colour changes over a very wide range of change in concentration of H+ ions. Question. Why does HCl(g) not conduct electricity when dissolved in toluene? Answer: It is because HCl cannot ionise in organic solvent like toluene. Short Questions: Question of A (b) solution of B? Answer: pH of pure water whereas pH of B will be more than that of pure water as it is a soluble base. Question: What happens when a solution of an acid is mixed with a solution of a base in a test tube? Answer: The salt formation takes place along with water when a solution of an acid is mixed with a solution of a base. Question: What happens when a solution of a base in a test tube? Answer: The salt formation takes place along with water when a solution of an acid is mixed with a solution of a base. Question: cake: (A) How will it affect the taste of the cake and why? (B) How can baking soda? Answer: (A) Baking soda be converted into baking soda be converted into baking soda is sodium hydrogen carbonate which decomposes to sodium carbonate, water and carbon dioxide on heating. Baking powder is a mixture of sodium hydrogen carbonate with tartaric acid, which readily reacts with sodium carbonate and neutralises it. Question X turns orange whereas solution X turns red, Which of these is a stronger acid? Answer: The universal indicator produces red colour for the strongest acid and orange colour for slightly weaker acid. As solution X turns orange whereas solution Y turns red, Y is a stronger acid than X.Question : What is the action of calcium oxide is a fairly strong base, and hydrochloric acid is a strong acid. The two substances readily reacts with and neutralise each other, forming the soluble salt calcium chloride (CaCl2) and water.CaO + 2HCl (dil) - CaCl2 + H2O Question: Answer the following:(A) How is tooth decay related to pH? How can it be prevented?(B) What is the change in colour of pH paper dipped in a solution having a pH = 13? Answer: (A) When we eat food containing sugar, then the bacteria present in our mouth break down the sugar to form acids such as lactic acid. This acid lowers the pH in the mouth making it acidic. Tooth decay starts when the pH of acid formed in the mouth falls below 5.5. This is because then the acid becomes strong enough to attack the enamel of our teeth and corrode it. This sets in tooth decay starts when the pH of acid formed in the mouth falls below 5.5. This is because then the acid becomes strong enough to acid formed in the mouth falls below 5.5. mouth thoroughly after eating food by rinsing it with lots of clean water. Many tooth pastes contain bases to neutralise the mouth acid. The pH of tooth paste is about 8.0. Therefore, using the tooth paste, which is generally basic, for cleaning the tooth paste is about 8.0. Therefore, using the tooth 8.0. The tooth paste is about 8.0. Therefore, using the tooth 8.0. The too (13) will have minimum hydrogen ion concentration i.e. alkaline solution and so will change colour of pH paper to light violet colour. Ouestion: A vellow powder X gives a pungent smell if left open in air. It is prepared by the reaction of dry compound Y with chlorine gas. It is used for disinfecting drinking water. Identify X and Y.Answer: The vellow powder X is bleaching powder which is prepared by the reaction of chlorine on dry slaked lime [Ca(OH)2]. Question: The tooth enamel is made of calcium phosphate [Ca3(PO4)2]. It is basic in nature. It starts getting corroded when the pH in the mouth is slightly acidic. This is the reason why toothpaste contain bases to neutralise the acid in the mouth. Question : Equal lengths of Mg ribbon are taken in test tube A. While acetic acid is added to test tube B. In which case the reaction would occur more vigorously and why? Answer: The reaction will take place more vigorously in test tube A as hydrochloric acid is a stronger acid than acetic acid. Ouestion: What do you understand by olfactory indicators? Answer: The substances whose smell changes in acidic or basic solutions are called olfactory indicators. An olfactory indicator usually works on the principle that when an acid or base is added to it, it loses its characteristic smell. Onion and vanilla extract are examples of olfactory indicators. Question : Farmers are using a large number of pesticides and fertilisers in their fields to increase crop production and to enhance their profits. But by doing so, they are causing damage to the soil as well as to the environment. Do you agree with this statement? Why should we avoid eating fruits and vegetables without washing them properly? Answer: Plants require a specific pH range for their healthy growth. By using a large number of pesticides and fertilizers, pH of the soil becomes infertile. This leads to soil erosion causing damage to the environment also. So, the use of these pesticides and fertilizers should be restricted. And these are toxic for us, that's why the fruits and vegetables must be properly washed before use. Ouestion: How do metal carbonate react with acids? Give one example. Answer: Metal carbonates react with acids to give salt, carbon dioxide + Watere.g. Na2CO3 + 2HCl \rightarrow 2NaCl + H2O + CO2 \uparrow Question: i. Hydrolysis of a salt of a weak acid and a weak base. They give acidic, neutral or basic (or alkaline) solutions in water depending upon the nature of weak acid and a weak base from which salt is produced.e.g. Ammonium acetate (CH3COONH4) gives neutral solution on hydrolysis. e.g. NaCl, KCl, KNO3 NaNO3 etc.Question: Name two substances from daily life which contain acid and two substances which contain base. Question: What is the colour of CuSO4. 5H2O and CuSO4? Question: Name the acid present in guava? Question: What is Milk of Magnesia? Is it strong or mild base? Question: Name two substances which contain base. Question: What is the colour of CuSO4. The pH of a sample of a vegetable soup was found to be 6.5. How is this soup likely to be used?Question: On adding dil.HCl to copper oxide powder, the solution formed is blue-green. Predict the new compound formed which impacts a blue-green colour to the solution? Question: What is the cure for bee string? Why? Question: What is meant by 'water of crystallization' of a substance? Describe an activity to show that copper sulphate crystals contain water of crystallization. Question : A compound 'X' of sodium is completely used in kitchen for making crispy pakeras. It is also used for curing acidity in the stomach? Identify X. Write its chemical formulae. State the reaction which takes place when it is heating during cooking. Question: Name three products obtained on electrolysis of an aqueous solution of Sodium Chloride. Why this process is called chlor-alkali process? Question: A calcium compound which is a yellowish white powder is used in paper industry?Question: Three acidic solutions A,B,C have pH=0,3,5 respectively.a) Which solution has the least number of H+ ions?Question: Write word equations and the balance the equation and the equat is used in small amount in making bread and cake. It is used to make cake soft and spongy. An aqueous solution of baking soda to make cake and bread spongy and soft?b) How does it help in fire extinguisher?c) Is the pH value of baking soda lower than or higher than 7Question: State the name and function of the acid present in our stomach. What remedy would you suggest to a person suffering from indigestion, pain and irritation in the stomach? Name the main ingredient of this remedy would you suggest to a person suffering from indigestion, pain and irritation in the stomach? Name the main ingredient of this remedy would you suggest to a person suffering from indigestion, pain and irritation in the stomach? Name the main ingredient of this remedy would you suggest to a person suffering from indigestion, pain and irritation in the stomach? Name the main ingredient of this remedy would you suggest to a person suffering from indigestion, pain and irritation in the stomach? Name the main ingredient of this remedy would you suggest to a person suffering from indigestion, pain and irritation in the stomach? Name the main ingredient of this remedy would you suggest to a person suffering from indigestion, pain and irritation in the stomach? Name the main ingredient of this remedy would you suggest to a person suffering from indigestion. declared 21st February of every year to be celebrated as International Mother Language day to promote dissemination of Mother Language of all, create awareness of linguistic and cultural traditions and diversity across the world and to inspire solidarity... Evaluation of the Answer Books is done under a well-settled Policy. To ensure that the evaluation is error free, CBSE is taking several steps. After strictly following these steps, the result is prepared. Though, CBSE is having a well-settled system of assessment.... 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